

# Evaluation of wastewater reclamation technologies based on in vitro and in vivo bioassays

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# ABSTRACT

When municipal secondary effluent is used as the main supplementation water source for surface water bodies, its potential adverse ecological effects should not be neglected. The objective of this work was to investigate the effectiveness of several technologies, i.e. combination of coagulation and sand filtration (CSF), ultraviolet (UV) irradiation, chlorination, ozonation, ultrafiltration (UF) and reverse osmosis filtration (RO), on the removal of acute ecotoxicity, genotoxicity and retinoic acid receptor (RAR) agonist activity from the municipal secondary effluent. The effects of treated effluents on the development of Japanese medaka (Oryzias latipes) embryos were also evaluated. The secondary effluent exhibited a mutagenic effect on Salmonella typhimurium strain TA 1535/pSK1002, acute invertebrate toxicity to Daphnia magna, and weak RAR  $\alpha$  activity. RO and ozonation demonstrated remarkable removals of the genotoxic effect, acute toxicity and RAR activity from secondary effluent, while chlorination could elevate both genotoxicity and acute toxicity. CSF, UV, UF, chlorination as well as RO could decrease the 4-day mortality of medaka embryos and accordingly increase the hatching success rate, comparing with the secondary effluent. Ozonation at 4 mg/l and higher doses, however, elicited significantly higher 4-day mortality, leading to the reduction of the hatching success rate.

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# 1. Introduction

With its rapid economical development, China has experienced growing water crisis during the last two decades, both in terms of water scarcity and quality deterioration. Effluent from municipal sewage treatment plants (STPs) has recently been focused as a novel water resource because of its relatively stable quality and quantity properties. In some northern cities in China, which are in serious shortage of surface water, the use of reclaimed municipal wastewater as a major supplementation water source for rivers and lakes is considered to be an important countermeasure to the increasing water crisis. However, the secondary effluent has been found to cause some adverse aqua-ecological effects. Aguayo et al. (2004) found that the effluent from 7 investigated STPs all showed toxic effects to Daphnia magna, while some samples showed estrogenicity and teratogenicity. Dizer et al. (2002) reported that samples taken from a river accepting the secondary effluent caused genotoxic

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responses in the umu assay. The secondary effluent was also reported to induce significant reduction of hatching rates and increase of embryo lesions of exposed Japanese medaka (Oryzias latipes) embryo (Zha and Wang, 2005). One of the possible causative substance group is retinoids, which possess a chemical structure or functional properties similar to vitamin A, act as signaling molecules, and regulate many processes critical to early embryonic development (Sucov and Evans, 1995). Degitz et al. (2000) reported that 6.25 µg/l all trans retinoic acid could elicit higher teratological rate and mortality of exposed Xenopus laevis. Methoprene acid, a metabolic degradation product of methoprene, a pesticide, and organochlorine pesticides have been shown to bind to members of the retinoic acid receptors (RAR) (Harmon et al., 1995; Lemaire et al., 2005). Although up to now, it is not clear whether the reported increase of teratological rates of Japanese medaka embryos were related to retinoids in the secondary effluent, the above adverse ecological effects should be considered when the secondary effluent from the STPs is used as the main supplementation water source for surface water bodies.

Generally, almost all reclamation schemes have adopted some add-on technologies, among which the combination of coagulation and sand filtration (CSF) is most often introduced to the existing conventional secondary treatment, to upgrade the quality of reclaimed water. To reduce the pathogenetic risk, disinfection using chlorine is generally performed. However, the adverse effects of chlorination have caused concerns over the formation of hazardous disinfection byproducts (DBPs), and ultraviolet (UV) irradiation has been focused as a substitute for chlorination disinfection (Jolis et al., 2001). With the rapid development of membrane technologies, reclamation of municipal wastewater using ultrafiltation (UF)and reverse osmosis (RO) have become increasingly attractive (Bourgeous et al., 2001; Qin et al., 2005). UF has been known for its high efficiency in the removal of particles and some colloidal organic compounds (Bian et al., 1999; Abdessemed et al., 2000). RO is a very promising technique because it removes the majority of compounds, both organic and inorganic, with a high efficiency (Qin et al., 2005). Ozone with its strong oxidation potential is effective in disinfection, decoloration, and decomposition of organic compounds. Takanashi et al. (2002) found that ozone treatment was effective for the removal of mutagen precursors from wastewater. Petala et al. (2006) have compared the performances of different coagulants on the removal of ecotoxic and mutagenic effects from the reclaimed secondary effluent. However, systematic comparison of the existing add-on treatment technologies has not yet been carried out with respect to their respective performance in the reduction of various ecological toxicities.

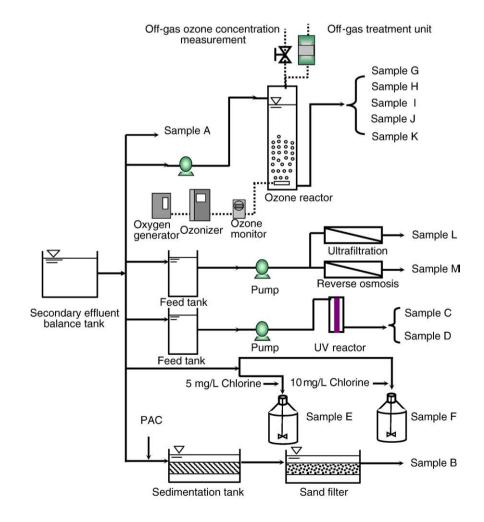


Fig. 1-Schematic diagram of reclamation treatment process and sampling.

In this study, the acute toxicity, genotoxicity and RAR activity in the reclaimed municipal effluent were respectively evaluated using *D. magna, umu* test and yeast two-hybrid test, and the performance of several water treatment technologies, namely CSF, chlorination, UV, ozonation, UF and RO, was evaluated for their respective removal of the ecological toxicities from the reclaimed wastewater. The effects of treated effluents on the development of Japanese medaka embryos were also evaluated. The results of this study will be useful for the selection of suitable treatment technologies to reclaim the secondary sewage effluent for replenishment of surface water.

# 2. Materials and methods

#### 2.1. Water samples and chemicals

Secondary municipal effluent was collected from an STP in Beijing, which has a treatment capacity of 1 million m<sup>3</sup> per day with a treatment train consisting of anoxic and aerobic treatments. In this STP, 10,000 m<sup>3</sup> per day secondary effluent was further treated with a CSF process for reuse. 100-1 secondary effluent was taken from a balance tank for chlorination, UV, UF and RO treatments in laboratory and 10-1 CSF samples were taken directly from the wastewater reclamation facility (Fig. 1). Ozonation experiments were conducted using an on-site treatment system. All samples were taken on the same day, and transferred to the laboratory and kept in 4 °C for further treatments or analyses.

Analytical standards for monochloroacetic acid (MCAA), monobromoacetic acid (MBAA), dichloroacetic acid (DCAA), dibromoacetic acid (DBAA) and trichloroacetic acid (TCAA) were purchased from Acros Organics (Belgium), and chloroform, bromodichloromethane, dibromochloromethane bromoform and 1,2-dibromopropane standards were obtained from Accustandard<sup>®</sup> Inc (USA). All of the chemical reagents used were of analytical grade if not noted specially.

#### 2.2. Treatment experiments

#### 2.2.1. Chlorination

A stock chlorine solution was prepared by diluting sodium hypochlorite with pure water to give a free chlorine concentration of 57 g/l. A given volume of the stock chlorine solution was added to 10 l secondary effluent in a sealed stainless vessel under continuous mixing for 24 h at ambient temperature. 0.025 N sodium sulfite was then added to the solution to eliminate residual chlorine. Free chlorine and total chlorine were measured using "Hach Colorimeter <sup>™</sup>II test kit for chlorine" (Hach, USA).

After Daphnia tests indicated that the acute toxicity increased following chlorination, chlorination on filtrated secondary effluent was again performed to determine the levels of DBPs formed at each chlorine dose.

## 2.2.2. Ozonation

A transparent polyvinyl chloride contact column ( $Ø200 \times 3$  m) with an effective volume of 78.5 l was used as ozonation reactor.

An ozonizer (OS-1N, Mitsubishi Electric Co. Japan) with a rated output of 2 g/h was used. The secondary effluent was continuously pumped into the contact column. Different ozone doses were obtained by adjusting voltage of the ozonizer. The hydraulic residence time (HRT) of the ozone treatment system was 20 min, and samples were taken after the system was operated stably for more than 1 h under each condition.

Ozone concentration of feed gas was monitored using a UV ozone monitor (Hare EG-600, Ebara Jitsugyo, Japan). The residual ozone in the vent gas was adsorbed by KI adsorption, and was then determined with the sodium thiosulfate titration method together with the dissolved ozone. The ozone dose on consumption base ( $C_C$ ) was calculated by subtracting the ozone amount in the vent gas and water from that in the feed gas as Eq. (1),

$$C_{\rm C} = \frac{C_{\rm F} \times F_{\rm G} - Q_{\rm O}}{F_{\rm W}} - C_{\rm D} \tag{1}$$

where  $C_{\rm C}$ =ozone dose on consumption base (mg/l),  $C_{\rm F}$ =inlet ozone concentration (mg/l),  $F_{\rm G}$ =flowrate of ozonized gas (l/min),  $Q_{\rm O}$ =residual ozone in vent gas (mg/min),  $C_{\rm D}$ =dissolved ozone concentration in water (mg/l) and  $F_{\rm w}$ =flowrate of water (l/min). The ozone dosage mentioned in this paper stands for  $C_{\rm C}$ .

## 2.2.3. Membrane filtration

The membrane filtration experiments were conducted by a flat sheet membrane apparatus (C10-T, Nitto Denko Matex, Japan) using UF membrane (50KD, Nitto Denko Matex, Japan) or RO membrane (Salt rejection 90.0%, Nitto Denko Matex, Japan). 8 l permeate was obtained through UF and RO filtration respectively from each 20 l secondary wastewater. The working pressure was maintained in the range of 0.3~0.5 MPa and the flowrate was approximately 0.7 l/min.

## 2.2.4. UV treatment

intensity.

The secondary wastewater was pumped into a stainless steel UV column reactor with effective volume of 160 ml and UV lamp intensity of 3.949 mw/cm<sup>2</sup>. Different UV irradiation doses were obtained by altering the HRT in the UV reactor (Tables 1 and 2).

## 2.3. Physico-chemical analysis

Samples were filtered through a 0.45 µm membrane for analyses. The DOC was determined by Total Organic Carbon Analyzer (Phoenix, 8000, Tekmar-Dohrmann Co. USA), and UV absorbance at 254 nm (UV254) by UV3100 (Hitachi Ltd. Japan). COD<sub>Mn</sub> was determined according to the Standard Methods of Water and Wastewater Monitoring of China, and ammonia

Table 1 – Main parameters of UV treatment						
Sample		Reactor volume	HRT s	UV intensity mw/cm²	UV dose <sup>a</sup> mJ/cm <sup>2</sup>	
С	1	160 mL	10	3.949	40	
D 0.4 160 mL 24 3.949 95 <sup>a</sup> UV dose was calculated by multiplying the HRT by the UV						

	Samples	COD <sub>Mn</sub>		DOC		UV254		NH <sub>4</sub> -N	
		mg/l	%	mg/l	%	$m^{-1}$	%	mg/l	%
А	Secondary effluent	7.0	-	7.9	-	15.2	-	1.9	-
В	CSF	а		6.4	19.0	13.7	9.9	1.7	10.5
С	UV40 mJ/cm <sup>2</sup>	а		7.7	2.0	13.0	14.0	1.9	0.0
D	UV94 mJ/cm <sup>2</sup>	а		7.6	3.0	12.8	15.4	1.9	0.0
E	Chlorination 5 mg/l	5.8	17.1	7.6	3.8	13.3	12.5	1.9	0.0
F	Chlorination 10 mg/l	5.5	21.7	7.1	9.3	11.9	21.8	1.3	31.6
G	Ozonation 2 mg/l	5.6	19.5	6.4	18.2	11.7	22.9	1.9	0.0
Н	Ozonation 3.8 mg/l	5.3	23.7	6.1	22.7	10.0	34.3	1.9	0.0
Ι	Ozonation 8.5 mg/l	4.9	29.3	6.0	23.6	6.0	60.2	2.1	-10.5
J	Ozonation 11.1 mg/l	4.8	31.6	5.7	27.8	4.7	68.8	2.3	-21.1
К	Ozonation 15 mg/l	4.5	35.2	5.5	30.6	4.9	67.5	2.5	-33.7
L	UF50K	а		6.2	20.8	11.7	23.1	1.1	42.1
М	RO	а		0.9	88.2	0.2	99.0	0.1	94.7

was determined by colorimetry using the nesslerization method.

### 2.4. Chlorination DBPs analysis

Trihalomethanes (THMs) and Haloacetic acids (HAAs) analysis were conducted using an Agilent 6890 N Gas Chromatograph (USA) equipped with an HP-5 fused silica capillary column (0.25 mm×30 m) and an electron capture detector (ECD). Four THMs, i.e., chloroform, bromodichloromethane, dibromochloromethane and bromoform were analyzed following liquid/liquid extraction with hexane (HPLC grade, Fisher Chemicals, USA), in accordance with U.S.EPA (1990) method 551.1. Five HAAs, i.e., MCAA, MBAA, DCAA, DBAA and TCAA, were analyzed using 1,2dibromopropane as an internal standard, following liquid/liquid extraction with methyl-t-butyl-ether (MTBE, HPLC grade, Fisher Chemicals, USA) and derivatization with acidic methanol (HPLC grade, Fisher Chemicals, USA) according to U.S.EPA (2003) method 552.3.

#### 2.5. Concentration of wastewater samples

Water samples before and after treatment were filtered through GF/C membrane (Whatman, UK) without pH adjustment, and then passed through Oasis HLB solid extraction cartridges (6 cc 500 mg, Waters, USA). The cartridges had been conditioned beforehand with 10 ml of methanol and 10 ml of distilled water. 2 l of each sample was loaded to one cartridge, then dried under a nitrogen flow and eluted with 6 ml of methanol. The eluate was dried under nitrogen flow. The dried residues were reconstituted with 200  $\mu$ l dimethylsulfoxide (DMSO, Ameresco, USA) and stored in the dark at –20 °C before used for the *umu* and yeast two-hybrid tests.

## 2.6. umu assay

The genotoxic effects of concentrated samples were determined with the SOS/*umu* bioassay (ISO 13829, 2000; Oda et al., 1985) using Salmonella typhimurium strain TA 1535/pSK1002. The test strain was provided by the Osaka Prefectural Institute of Public Health, Japan. In this strain, the multicopy plasmid pSK 1002 bearing an *umuC/lacZ* gene fusion product was introduced into S. typhimurium TA1535, and the *umu* operon was genetically regulated by the SOS genes *recA* and *lexA*. The detailed test procedure was performed as previously described by Hu et al. (2007). In this assay, 4-nitroquinoline oxide (4-NQO) was used as positive control and DMSO as negative control.

## 2.7. Yeast two-hybrid assay for RAR agonist activity

The yeast two-hybrid assay was used to test the RAR  $\alpha$  agonist activity. The ligand-binding domain of nuclear receptor of RAR $\alpha$  was cloned by RT–PCR from human mRNA. These genes were subcloned into pGBT9, so that they were in the same translational reading frame as the vector's GAL4DNA binding domain. pGBT9-NRs and pGAD424-TIF2 were integrated into *Saccharomyces cerevisiae* Y190, which was provided by Osaka University. The assay procedure was the same as described in a previous paper (Nishikawa et al., 1999).

#### 2.8. D. magna test

D. magna bioassay was carried out using dormant eggs and the salts for preparation of standard freshwater (ISO formula) contained in the "Daphtoxkit F<sup>™</sup>" (Microtests Inc, Belgium). Hatching of ephippia and preparation of standard freshwater were performed according to the manufacturer's instructions. The ephippia were transferred to hatching Petri dishes with 50 ml pre-aerated standard freshwater, thereafter covered and incubated for 72 h, at 20–22 °C under continuous illumination of 6000 lx. Subsequently, the neonates were feed and transferred to standard freshwater with a Pasteur pipette 2 h before bioassays.

A dilution series of treated and untreated water samples was prepared by serial 1:1 dilution with standard freshwater. Assays were carried out in 24-well plates. Five neonates were transferred into each well, which contained 10 ml water sample. Freshwater controls were included in every test. Tests were performed in quadruplicate. The plates were covered and incubated at 20 °C in the dark. After 24 h and 48 h incubation, the number of dead and immobilized neonates was recorded and the percent mortality was calculated.

## 2.9. Japanese medaka (O. latipes) embryo exposure test

Eggs were collected daily from the breeding females at less than 2 h after fertilization. Individual egg was carefully detached from the clusters using sucker and then inspected for fertilization using stereoscopic microscope. Eggs with the migration of oil globules to the vegetal pole were deemed to be fertilized (Kirchen and West, 1976).

Medaka embryos collected at less than 4 h after fertilization were exposed to the treated and untreated water samples without dilution. For each sample, 100 embryos were randomly separated into two groups with 50 embryos in each glass Petri dish (Ø15 cm×2 cm) containing 100 ml water sample, and then incubated at 25 °C under a 16:8 h light:dark photoperiod. The water samples were renewed every single day until all the able embryos hatched. Larvae were fed with newly hatched brine shrimp once a day. The embryos and larvae were examined daily for mortality under a stereoscopic microscope, and the dead embryos were discarded. Embryos cultured in tap water dechlorinated with granular activated carbon were used as control.

# 3. Results and discussion

#### 3.1. Water quality

Water quality parameters before and after treatments are presented in Table 2. It is clear that RO filtration could significantly improve water quality with 88.2% of DOC, 94.7% of ammonia and 99% of UV254 removals from the secondary effluent. The DOC and UV254 removals of ozonation at a dose of 2 mg/l (18.2% and 22.9%) were equivalent to those of UF treatment (20.8% and 23.1%). The increase of ozone dose from 2 mg/l to 8.5 mg/l led to the increase of UV254 removal to 60.2%. The DOC removal, however, was not improved so markedly. The CSF treatment could remove 19.0% of DOC and 9.9% of UV254. UV irradiation and chlorination were not so effective for DOC removal (<10%), but could remove a little UV254 (from 14.0% to 21.8% according to doses). Chlorination at a chlorine dose of 10 mg/l led to a removal of ammonia of 31.6%, which should be caused by break point reaction.

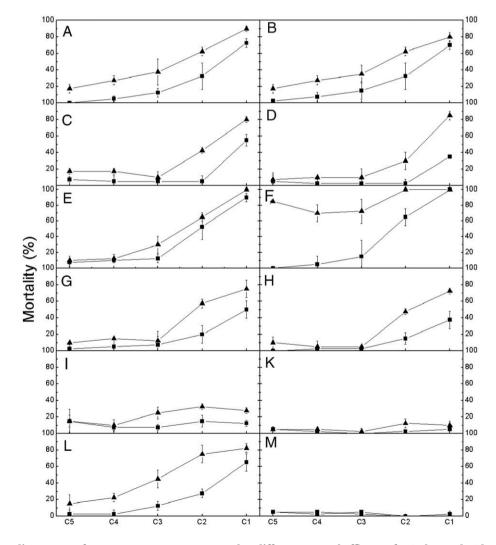


Fig. 2–Dose–mortality curves of D. magana neonates exposed to different treated effluents for 24 h  $\blacksquare$  and 48 h  $\blacktriangle$ . C1, C2, C3, C4, and C5 represent 100%, 50%, 25%, 12.5%, and 6.25% of the effluents diluted with standard freshwater. The data are percent mortalities corrected for the natural mortality; values are the means ± SE of 3 independent experiments, each with four replicates.

#### 3.2. Acute invertebrate toxicity

Fig. 2A shows that there exists a clear dose-response effect between the fraction of the secondary effluent and the mortality of neonates, and that the mortality of neonates expose to the secondary effluent without dilution for 24 h and 48 h was 72.5% and 90.0%, respectively, indicating the high acute invertebrate toxicity of the secondary effluent to neonates. Ozonation was effective for the reduction of acute toxicity from the secondary effluent. The 48-h neonate mortality was almost completely removed for the samples diluted by 4 times at an ozone dose of 4 mg/l (Fig. 2H). The dose-response effect between the fraction of the secondary effluent and the mortality of neonates almost disappeared at an ozone dose of 8 mg/l (Fig. 2I), and ozone dose of 15 mg/l (Fig. 2K) reduced the mortality nearly to blank level, suggesting that such an ozone dose might be sufficient for the complete removal of the acute toxicity from the secondary effluent.

It is clear that RO treatment was very effective for the removal of acute toxicity from the secondary effluent (Fig. 2M). UV treatment resulted in a minor decrease of 24-h mortality while little effect was observed on 48-h mortality, which was in accordance with the effect of ozonation at an ozone dose of 2 mg/l (Fig. 2C, D and G). It is possible that the UV254 removal by UV irradiation might be related with the reduction of acute toxicity. On the other hand, the mortality of neonates was not perceptibly improved by the treatments of CSF and UF although these two treatments could also remove DOC and UV254 to some extent. It is possible that the removed compounds in these two processes might be different from those in UV and ozone processes.

It is of concern that the mortality of neonates exposed to the chlorinated secondary effluent increased in spite of the moderate removal of UV254. The 24-h and 48-h mortalities of the secondary effluent increased from the approximately 80% and 90% to 90% and 100% after chlorination at a 5 mg/l chlorine dose. Chlorination at 10 mg/l led to a mortality of 100% for both the 24-h and 48-h cultivation. It should be noted that the 48-h mortality of neonates in system F was as high as 70% even when the chlorinated water was diluted by 4 times. Since dechlorination was performed after the chlorination treatment, the acute toxicity should not be originated from the residual chlorine or chloramines. It is possible that the

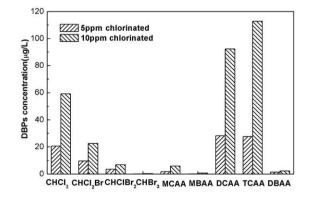


Fig. 3 – THMs and HAAs measured in the chlorinated secondary effluent.

Table 3 – Acute toxicity values of potential DBPs for D. magna					
Potential chlorination DBPs	test	Value (mg/l)	Reference		
Trichloromethane	48 h-LC50	229.21	Genoni, 1997		
Trichloromethane	24 h-LC50	64.23	Guilhermino et al., 2000		
Trichloromethane	24 h-EC50	573.014	Guilhermino et al., 2000		
1,2-Dichloropropane	48 h-LC50	49.659	Genoni, 1997		
1,2-Dichlorobenzene	48 h-LC50	3.528	Genoni, 1997		
2,4,5-Trichlorophenol	48 h-LC50	1.801	Genoni, 1997		
Chloroacetate	48 h-LC50	76.998	Genoni, 1997		
1,2,3-Trichloropropane	48 h-LC50	35.366	Genoni, 1997		
Formaldehyde	48 h-EC50	29	Janssen and		
Formaldehyde	24 h-EC50	57	Persoone, 1993 Janssen and Persoone, 1993		

increased acute toxicity was originated from the formation of some chlorination by-products. So, the DBP (4 THMs and 5 HAAs) formation abilities of the secondary effluent were evaluated, and the results are shown in Fig. 3. CHCl<sub>3</sub>, DCAA and TCAA were found to be the major ones among the 9 DBPs, and their concentrations increased markedly when the chlorine dose was increased from 5 mg/l to 10 mg/l. However, as shown in Table 3, the LC50 values of the conventional DBPs for D. magna are in mg/l levels, which are approximately 2 or 3 orders higher than the concentrations in the chlorinated samples. So, these conventional DBPs might not be the main cause for the increase of acute toxicity after chlorination. The secondary effluent is a very complex matrix, and its chlorinated DBPs have not yet been well understood. The formation of some unknown DBPs and the interactive effects among different DBPs might be responsible for the increased toxicities.

# 3.3. Genotoxicity

Genotoxicity of the secondary effluent before and after treatments was evaluated using S. typhimurium strain TA 1535r/pSK1002 without addition of metabolic activation system S9 mix, and is presented in Fig. 4. The genotoxic response induced by the secondary effluent was approximately 6 times

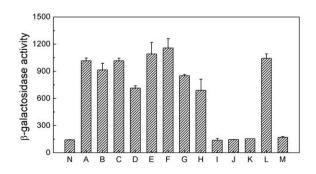


Fig. 4–Induced genotoxic activity of different treated effluents. N means the negative control; the values are the means ± SE of 3 experiments.

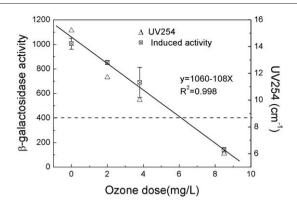


Fig. 5 – Induced mutagenic activity and UV 254 under different ozone dose. Dash line means the positive value of the *umu* test.

higher than that of the negative control, indicating that the secondary effluent had a relatively high genotoxicity. It is clear that ozonation and RO were the only two methods that are effective for the reduction of genotoxicity from the secondary effluent. Ozonation at an ozone dose of 8.5 mg/l and the RO treatment could almost reduce the genotoxic activity to the negative control level. The evolution of genotoxic activity is plotted against ozone dose together with UV254 (Fig. 5). It is clear that the induced activity was linearly reduced together with UV254 with the increase of ozone dose. According to this plot, it is speculated that an ozone dose of 5.9 mg/l could reduce the genotoxic potential from positive to negative. The above results suggested that the genotoxicity in the secondary effluent might be related with presence of some unsaturated compounds.

On the other hand, the CSF, UF with 50000D and UV irradiation at 40 mj/cm<sup>2</sup> did not perceptibly change the genotoxic potential of the secondary effluent; while UV irradiation at 95 mj/cm<sup>2</sup> could remove 29.6% genotoxic potential from the secondary effluent (Fig. 4). UV irradiation might have broken some unsaturated structures of organic compounds which possessed genotoxicity. Although chlorination led to the decrease of UV254, the genotoxic potential increased by 8.7% for 5 mg/l chlorine dose and 15.2% for 10 mg/l chlorine dose, respectively. The existence of ammonia in the

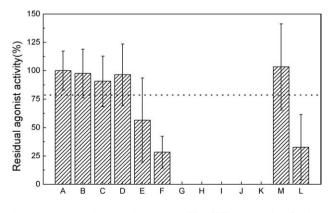


Fig. 6 – RAR agonist activity removal by different technologies. The values are the means ± SE of 3 experiments.

secondary effluent caused the formation of chloramines during chlorination, which may produce DBPs with toxicities far more potent than those currently regulated (Plewa et al., 2004; Choi and Valentine, 2002; Richardson et al., 2007). Further works are necessary to elucidate the reason for the increase of genetic toxicity.

# 3.4. RAR $\alpha$ agonist activity

The RAR $\alpha$  activities of the secondary effluent as well as treated effluents were determined. The secondary effluent showed weak agonist activity, and the all trans retinoic acid equivalent was 13.4±2.7 ng/l. Degitz et al. (2000) reported that 6.25 µg/l all trans retinoic acid could elicit higher teratological rates and mortality of exposed X. *laevis*. However, research regarding the agonist activities in the secondary effluent has been very limited. Fig. 6 shows the RAR $\alpha$  agonist activity removal performance of different reclamation technologies. UV, CSF and UF could not remarkably remove the RAR $\alpha$  activity, while chlorination treatment could remove the activity to some extent. Ozonation was very effective to reduce the RAR $\alpha$ agonist activity, and the activity was almost completely removed even at an ozone dose of 2 mg/l.

#### 3.5. Effects on Japanese medaka embryos development

As shown in Fig. 7, the 4-day mortality in the secondary effluent was 22%, which was effectively promoted through RO treatment to a level equivalent to that of control. The 4-day mortality remained almost unchanged for CSF, UV and UF treatments. Again, the 4-day mortality increased to 32% for chlorination at 10 mg/l. Surprisingly, ozonation remarkably increased the mortality: the mortality was 32% for ozonation at 2 mg/l, which increased to 52% at 4 mg/l. But further increase of ozone dose did not lead to the increase of mortality.

Abnormalities of embryos were observed from the fourth day (Fig. 7). The lesions would, to some extent, affect the hatching success of embryos and the growth and survival of the larvae (Kjørsvik et al., 1990). Absolute majority of embryos were normal in the controls and RO effluent (Fig. 8a), in which

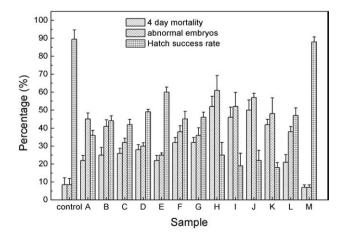


Fig. 7–The effects of different treatments on Medaka embryos development. Values are the means±SE of 2 independent experiments, each with 50 embryos.



Fig. 8 – Teratological development forms of embryos: (A) normal embryos; (B) lesion embryos with extruded vesicle; (C) lesion embryos with hemorrhage; (D) head-first hatch out pattern; (E) deformed larvae with impaired spine; (F) deformed larvae with cranial herniation.

all abnormal eggs died before the fourth day and no more was observed later. The secondary effluent elicited several morphologic effects on embryos as well as the effluents of other treatment technologies as shown in Fig. 8. Most abnormal embryos emerging as a vesicle extruding outside of yolk sphere wall (Fig. 8b) ended up with death. Internal hemorrhage (Fig. 8c), which was the rare case and mainly occurred in the secondary effluent and chlorinated samples, did not statistically interfere with hatching success (Table 4). In Fig. 8d cases, the embryos did not hatch out in the normal tail-first pattern, but instead hatched out in an abnormal head-first pattern. These embryos showed progressing embryonic development but most failed to hatch. Although some embryos survived the hatching process, they were

Sample	Teratological development form of the embryos					
	Hemorrhage	0	Head-first hatch out			
А	+	++	+	-		
В	+	++	+	-		
С	-	+	+	-		
D	-	+	+	-		
Е	-	+	+	-		
F	+	++	+	-		
G	-	+	+	-		
Н	-	+	+	-		
Ι	-	+	+	-		
J	-	+	++	+		
K	-	+	++	+		
L	-	+	-	-		
М	-	-	-	-		

unable to survive for more than 2 days due to their greatly impaired swimming ability (Fig. 8e). The case of Fig. 8d was frequently observed in ozone treated effluents, which consequently leaded to the decrease of hatching success rate. A more severe developmental effect on embryos showing cranial herniation (Fig. 8f) was observed in high dose ozone treated water. It is not clear whether the cranial herniation resulted from skull development defects or impairment during the abnormal head-first hatching process.

The hatching rate, which is an important parameter for indicating the quality of larvae and the survival of the target species, is sensitive to changes of the environment (Detlaf et al., 1993). The hatching success rate of embryos exposed to the secondary effluent was 36%, which was significantly lower than that of the controls (89.5%). Comparing with the secondary effluent, CSF, UV, chlorination, UF and ozonation at 2 mg/l could elevate the hatching success rate by 10% or so. However, the increase of ozone dose led to the decrease of the hatching rate significantly. Among all of the treatment technologies, only RO could elevate the hatching success rate of embryos nearly to the level of the controls. Grotmol and Totland (2000) found that the survival rate of Hippoglossus hippoglossus embryos increased after treated with 4 mg/l ozone for 0.5 min for disinfection, but decreased under higher ozone doses. In this work, residual ozone was erased before exposing the eggs. Thus, it is speculated that some byproducts formed during ozonation caused the decrease of hatching success rate. Organic ozonation byproducts reported include aldehydes, ketones, ketoaldehydes, carboxylic acids, aldo acids, keto acids, hydroxy acids, alcohols, esters, and alkanes (Coleman et al., 1992; Killops, 1986). It is speculated that these ozonation byproducts with high polar and small molecular weight might have easier access into the embryos, disturbing the normal development of embryos. Further studies are required to clarify the ozonation byproducts responsible for the teratogenic development of embryos, and to find an effective abatement method for the removal of the possible byproducts. As small molecular ozonation byproducts are usually biodegradable, it is possible that a subsequent biofilter might be able to remove these compounds.

# 4. Conclusion

Two in vitro assay and two in vivo assay showed that the secondary effluent have genotoxicity, weak RAR  $\alpha$  activity, acute invertebrate toxicity as well as remarkably adverse effect on medaka embryos development. RO is the most effective technologies to remove most toxicants from the secondary effluent. CFS and UF showed little effect on the genotoxicity,  $RAR\alpha$  activity and acute invertebrate toxicity, while UV elicited abatement of genotoxicity only at a high dose. Ozonation was found to be effective to remove the genotoxicity, RARa agonist activity and acute invertebrate toxicity, but the ozonated effluent at higher than 4 mg/l would elicit the significant decrease of hatching success of medaka embryos, which requires a subsequent treatment to remove the bio-available byproducts. Increase of genotoxicity and acute invertebrate toxicity were observed in chlorinated effluents, which might be caused by the production of some chlorinated byproducts.

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